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## Structure Reports

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## 5-(3-Methyl-5-phenoxy-1-phenylpyrazol-4-yl)-1,3,4-thiadiazol-2-amine

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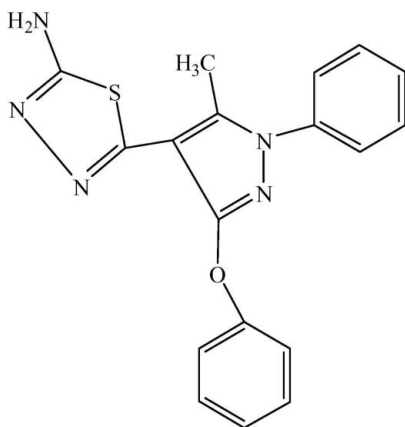
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Key indicators: single-crystal X-ray study;  $T = 153$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å;  $R$  factor = 0.036;  $wR$  factor = 0.101; data-to-parameter ratio = 17.2.

There are two molecules in the asymmetric unit of the title compound,  $\text{C}_{18}\text{H}_{15}\text{N}_5\text{OS}$ . The two molecules differ by less than  $3^\circ$  in the dihedral angles between corresponding planar units. Hydrogen bonding between the two molecules creates a one-dimensional chain along the  $c$  direction.

## Related literature

For related literature, see: Akbas *et al.* (2005); Daidone *et al.* (2004); Foroumadi *et al.* (2002); Park *et al.* (2005); Supuran & Scozzafava (2000); Thomasco *et al.* (2003).



## Experimental

## Crystal data

$\text{C}_{18}\text{H}_{15}\text{N}_5\text{OS}$   
 $M_r = 349.41$   
 Monoclinic,  $P2_1/c$   
 $a = 20.2987$  (5) Å

$b = 9.6934$  (3) Å  
 $c = 19.0726$  (6) Å  
 $\beta = 114.846$  (1)°  
 $V = 3405.43$  (17) Å<sup>3</sup>

$Z = 8$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.21$  mm<sup>-1</sup>

$T = 153$  (2) K  
 $0.66 \times 0.43 \times 0.31$  mm

## Data collection

Rigaku R-Axis SPIDER  
 diffractometer  
 Absorption correction: multi-scan  
 (ABSCOR; Higashi, 1995)  
 $T_{\min} = 0.876$ ,  $T_{\max} = 0.939$

32693 measured reflections  
 7772 independent reflections  
 6769 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.020$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.036$   
 $wR(F^2) = 0.101$   
 $S = 1.05$   
 7772 reflections

453 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\max} = 0.24$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.41$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H1A}\cdots\text{N7}$	0.88	2.19	3.0266 (16)	158
$\text{N1}-\text{H1B}\cdots\text{N4}^i$	0.88	2.21	3.0113 (15)	150
$\text{N6}-\text{H6A}\cdots\text{N2}$	0.88	2.11	2.9467 (15)	159
$\text{N6}-\text{H6B}\cdots\text{N9}^{ii}$	0.88	2.23	2.9915 (16)	145

Symmetry codes: (i)  $x, -y + \frac{3}{2}, z + \frac{1}{2}$ ; (ii)  $x, -y + \frac{3}{2}, z - \frac{1}{2}$ .

Data collection: *RAPID-AUTO* (Rigaku, 2004); cell refinement: *RAPID-AUTO*; data reduction: *RAPID-AUTO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: OM2139).

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**supplementary materials**

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## 5-(3-Methyl-5-phenoxy-1-phenylpyrazol-4-yl)-1,3,4-thiadiazol-2-amine

Y.-R. Sun, G. Liu, C.-J. Liu and J.-D. Wang

### Comment

Compounds including 1,3,4-thiadiazole ring are known to possess several biological properties, such as antibacterial, anti-tubercular, antineoplastic activities (Thomasco *et al.*, 2003; Foroumadi *et al.*, 2002; Supuran & Scozzafava, 2000). Pyrazoles also have widespread biological activity, such as antiseptics, antileukosis, antitumor (Akbas *et al.*, 2005; Daidone *et al.*, 2004; Park *et al.*, 2005). In an attempt to achieve new compounds with possible biological activities, we designed and synthesized 2-amino-5-(1-phenyl-3-methyl-5-phenoxy-pyrazole-4-yl)-1,3,4-thiadiazole(1). The asymmetric unit has two molecules, and structures of the two molecules are nearly identical.

### Experimental

To the mixture of 1-phenyl-3-methyl-5-phenoxy-pyrazole-4-formylic acid(4 mmol) and aminothiourea(4 mmol), phosphorus oxychloride(10 mL) was slowly added with stirring in ice bath. The reaction mixture was slowly heated to reflux 2 h at about 358 K, then cooled, and cold water (30 mL) added. The solution was slowly heated to reflux 6 h at about 393 K, then filtered. To neutralize the filtrate, 25% KOH was added to reach pH = 8–9. The product was isolated by filtration, washed with water, and dried at room temperature for a yield 65.2% (0.912 g), m.p. 497–498 K. Block-like single-crystal of the title compound were grown from a solution of ethanol by slow evaporation.

### Refinement

The H atoms attached to the parent C or N atoms were geometrically fixed, and were treated as riding atoms, whereas the methyl H atoms were refined as rigid groups, coupled with rotation of the group to best fit the electron density map. The distances were N—H = 0.88 Å, C—H = 0.95–0.98 Å, with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C or N})$ ,  $U(\text{iso}(\text{methylH})) = 1.5U_{\text{eq}}(\text{C})$ .

### Figures

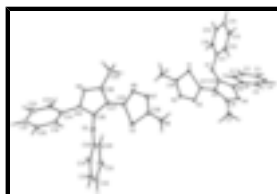


Fig. 1. A view of the molecular structure of the two molecules of the asymmetric unit, showing the atom numbering scheme. Displacement ellipsoids are drawn at the 30% probability level, and H atoms are shown as small spheres of arbitrary radii.

## 5-(3-Methyl-5-phenoxy-1-phenylpyrazol-4-yl)-1,3,4-thiadiazol-2-amine

### Crystal data

$\text{C}_{18}\text{H}_{15}\text{N}_5\text{OS}$

$M_r = 349.41$

$D_x = 1.363 \text{ Mg m}^{-3}$

Melting point: 498 K

# supplementary materials

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Monoclinic,  $P2_1/c$

$a = 20.2987$  (5) Å

$b = 9.6934$  (3) Å

$c = 19.0726$  (6) Å

$\beta = 114.846$  (1)°

$V = 3405.43$  (17) Å<sup>3</sup>

$Z = 8$

$F_{000} = 1456$

Mo  $K\alpha$  radiation

$\lambda = 0.71073$  Å

Cell parameters from 26663 reflections

$\theta = 3.0$ – $27.5$ °

$\mu = 0.21$  mm<sup>-1</sup>

$T = 153$  (2) K

Block, colorless

$0.66 \times 0.43 \times 0.31$  mm

## Data collection

Rigaku R-Axis SPIDER  
diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 153$ (2) K

$\omega$  scans

Absorption correction: empirical (using intensity  
measurements)

(ABSCOR; Higashi, 1995)

$T_{\min} = 0.876$ ,  $T_{\max} = 0.939$

32693 measured reflections

7772 independent reflections

6769 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.020$

$\theta_{\max} = 27.5$ °

$\theta_{\min} = 3.0$ °

$h = -24 \rightarrow 26$

$k = -12 \rightarrow 12$

$l = -24 \rightarrow 23$

## Refinement

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.036$

$wR(F^2) = 0.101$

$S = 1.05$

7772 reflections

453 parameters

Primary atom site location: structure-invariant direct  
methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring  
sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0577P)^2 + 0.8824P]$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 0.24$  e Å<sup>-3</sup>

$\Delta\rho_{\min} = -0.41$  e Å<sup>-3</sup>

Extinction correction: none

## Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculat-

ing  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.430303 (16)	0.79622 (3)	0.268570 (16)	0.02420 (8)
S2	0.053694 (17)	0.72302 (4)	0.263009 (18)	0.03485 (10)
O1	0.52710 (4)	0.88205 (9)	0.18168 (5)	0.02343 (18)
O2	-0.04172 (5)	0.82952 (9)	0.34655 (5)	0.02735 (19)
N1	0.33984 (6)	0.79945 (13)	0.34137 (6)	0.0323 (3)
H1A	0.2967	0.7916	0.3416	0.039*
H1B	0.3779	0.8193	0.3844	0.039*
N2	0.29285 (6)	0.75129 (12)	0.20940 (6)	0.0250 (2)
N3	0.31332 (5)	0.73761 (11)	0.14935 (6)	0.0236 (2)
N4	0.43532 (5)	0.68631 (11)	0.01202 (6)	0.0234 (2)
N5	0.49192 (5)	0.76583 (11)	0.06123 (6)	0.0225 (2)
N6	0.14095 (6)	0.71887 (15)	0.18704 (7)	0.0384 (3)
H6A	0.1835	0.7101	0.1857	0.046*
H6B	0.1017	0.7308	0.1439	0.046*
N7	0.19232 (6)	0.69641 (13)	0.32166 (6)	0.0295 (2)
N8	0.17298 (6)	0.68692 (12)	0.38289 (6)	0.0285 (2)
N9	0.05654 (5)	0.67127 (12)	0.52749 (6)	0.0269 (2)
N10	-0.00125 (5)	0.74338 (12)	0.47454 (6)	0.0249 (2)
C1	0.58095 (6)	0.93578 (13)	0.06074 (7)	0.0267 (3)
H1C	0.5691	0.9941	0.0938	0.032*
C2	0.63228 (7)	0.97676 (15)	0.03447 (8)	0.0316 (3)
H2B	0.6565	1.0627	0.0506	0.038*
C3	0.64825 (8)	0.89297 (16)	-0.01502 (9)	0.0380 (3)
H3B	0.6829	0.9219	-0.0333	0.046*
C4	0.61364 (9)	0.76660 (17)	-0.03796 (10)	0.0414 (3)
H4B	0.6242	0.7099	-0.0726	0.050*
C5	0.56374 (8)	0.72254 (15)	-0.01070 (8)	0.0334 (3)
H5A	0.5412	0.6348	-0.0251	0.040*
C6	0.54718 (6)	0.80857 (13)	0.03806 (7)	0.0235 (2)
C7	0.32015 (7)	0.59679 (14)	0.00612 (7)	0.0283 (3)
H7B	0.3224	0.5393	-0.0351	0.042*
H7C	0.2798	0.6622	-0.0161	0.042*
H7D	0.3126	0.5380	0.0439	0.042*
C8	0.38972 (6)	0.67421 (12)	0.04516 (6)	0.0218 (2)
C9	0.41636 (6)	0.74462 (12)	0.11713 (6)	0.0212 (2)
C10	0.48134 (6)	0.80017 (12)	0.12464 (6)	0.0209 (2)
C11	0.59244 (6)	0.82460 (12)	0.23525 (6)	0.0206 (2)
C12	0.60255 (7)	0.68515 (13)	0.25028 (8)	0.0289 (3)
H12A	0.5653	0.6208	0.2227	0.035*
C13	0.66881 (8)	0.64136 (15)	0.30706 (8)	0.0342 (3)
H13A	0.6770	0.5457	0.3182	0.041*
C14	0.72283 (7)	0.73517 (15)	0.34745 (8)	0.0324 (3)

## supplementary materials

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H14A	0.7677	0.7042	0.3864	0.039*
C15	0.71116 (7)	0.87414 (15)	0.33085 (8)	0.0334 (3)
H15A	0.7482	0.9389	0.3583	0.040*
C16	0.64566 (7)	0.91945 (13)	0.27442 (7)	0.0288 (3)
H16A	0.6376	1.0150	0.2629	0.035*
C17	0.38236 (6)	0.75766 (12)	0.17010 (6)	0.0208 (2)
C18	0.34753 (7)	0.78104 (12)	0.27530 (7)	0.0236 (2)
C19	-0.08683 (7)	0.92639 (14)	0.46842 (8)	0.0306 (3)
H19A	-0.0778	0.9722	0.4293	0.037*
C20	-0.13382 (7)	0.98335 (16)	0.49647 (8)	0.0371 (3)
H20A	-0.1574	1.0683	0.4760	0.044*
C21	-0.14653 (8)	0.9176 (2)	0.55371 (9)	0.0460 (4)
H21A	-0.1782	0.9578	0.5731	0.055*
C22	-0.11300 (8)	0.7928 (2)	0.58291 (9)	0.0484 (4)
H22A	-0.1219	0.7475	0.6223	0.058*
C23	-0.06649 (7)	0.73330 (17)	0.55515 (8)	0.0360 (3)
H23A	-0.0440	0.6471	0.5747	0.043*
C24	-0.05322 (6)	0.80151 (14)	0.49835 (7)	0.0261 (3)
C25	0.16923 (7)	0.56799 (15)	0.53478 (8)	0.0318 (3)
H25A	0.1701	0.5290	0.5826	0.048*
H25B	0.2101	0.6317	0.5473	0.048*
H25C	0.1731	0.4934	0.5020	0.048*
C26	0.09973 (6)	0.64382 (13)	0.49282 (7)	0.0248 (2)
C27	0.07052 (6)	0.69711 (13)	0.41642 (7)	0.0238 (2)
C28	0.00615 (6)	0.75820 (13)	0.40755 (7)	0.0238 (2)
C29	-0.10629 (6)	0.76360 (14)	0.29741 (7)	0.0249 (2)
C30	-0.11821 (7)	0.62435 (15)	0.29970 (9)	0.0356 (3)
H30A	-0.0826	0.5666	0.3365	0.043*
C31	-0.18401 (9)	0.57004 (19)	0.24642 (10)	0.0491 (4)
H31A	-0.1932	0.4741	0.2471	0.059*
C32	-0.23534 (8)	0.6529 (2)	0.19326 (9)	0.0497 (4)
H32A	-0.2797	0.6145	0.1571	0.060*
C33	-0.22249 (8)	0.7914 (2)	0.19243 (9)	0.0504 (5)
H33A	-0.2583	0.8489	0.1557	0.061*
C34	-0.15764 (8)	0.84882 (17)	0.24470 (8)	0.0378 (3)
H34A	-0.1489	0.9450	0.2441	0.045*
C35	0.10357 (6)	0.69930 (13)	0.36219 (7)	0.0242 (2)
C36	0.13585 (7)	0.71394 (14)	0.25529 (7)	0.0280 (3)

### Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
S1	0.02217 (15)	0.03248 (17)	0.01766 (15)	-0.00192 (11)	0.00807 (12)	-0.00262 (11)
S2	0.02132 (15)	0.0625 (2)	0.01953 (16)	0.00335 (14)	0.00740 (13)	0.00420 (14)
O1	0.0210 (4)	0.0248 (4)	0.0194 (4)	0.0012 (3)	0.0036 (3)	-0.0025 (3)
O2	0.0248 (4)	0.0307 (5)	0.0223 (4)	0.0007 (3)	0.0059 (4)	0.0038 (3)
N1	0.0266 (5)	0.0531 (7)	0.0187 (5)	0.0006 (5)	0.0112 (4)	-0.0021 (5)
N2	0.0228 (5)	0.0347 (6)	0.0186 (5)	0.0040 (4)	0.0098 (4)	0.0025 (4)

N3	0.0222 (5)	0.0309 (5)	0.0181 (5)	0.0034 (4)	0.0090 (4)	0.0021 (4)
N4	0.0203 (5)	0.0300 (5)	0.0181 (5)	-0.0027 (4)	0.0064 (4)	-0.0027 (4)
N5	0.0190 (5)	0.0291 (5)	0.0184 (5)	-0.0018 (4)	0.0070 (4)	-0.0017 (4)
N6	0.0264 (6)	0.0677 (9)	0.0219 (6)	0.0031 (5)	0.0110 (5)	0.0056 (5)
N7	0.0234 (5)	0.0446 (7)	0.0214 (5)	0.0003 (4)	0.0102 (4)	0.0019 (4)
N8	0.0228 (5)	0.0421 (6)	0.0203 (5)	-0.0014 (4)	0.0088 (4)	0.0003 (4)
N9	0.0213 (5)	0.0372 (6)	0.0192 (5)	0.0019 (4)	0.0058 (4)	0.0016 (4)
N10	0.0194 (5)	0.0349 (6)	0.0189 (5)	0.0014 (4)	0.0065 (4)	0.0007 (4)
C1	0.0248 (6)	0.0296 (6)	0.0249 (6)	0.0019 (5)	0.0094 (5)	0.0029 (5)
C2	0.0276 (6)	0.0338 (7)	0.0319 (7)	-0.0019 (5)	0.0111 (5)	0.0075 (5)
C3	0.0332 (7)	0.0480 (9)	0.0409 (8)	-0.0021 (6)	0.0235 (6)	0.0069 (6)
C4	0.0443 (8)	0.0474 (9)	0.0465 (9)	-0.0021 (7)	0.0328 (8)	-0.0055 (7)
C5	0.0348 (7)	0.0354 (7)	0.0379 (7)	-0.0032 (5)	0.0230 (6)	-0.0048 (6)
C6	0.0190 (5)	0.0310 (6)	0.0209 (6)	0.0022 (4)	0.0088 (5)	0.0045 (5)
C7	0.0250 (6)	0.0360 (7)	0.0230 (6)	-0.0064 (5)	0.0094 (5)	-0.0040 (5)
C8	0.0215 (5)	0.0261 (6)	0.0166 (5)	0.0001 (4)	0.0066 (4)	0.0007 (4)
C9	0.0213 (5)	0.0251 (6)	0.0165 (5)	0.0014 (4)	0.0071 (4)	0.0016 (4)
C10	0.0205 (5)	0.0236 (5)	0.0167 (5)	0.0020 (4)	0.0059 (4)	0.0006 (4)
C11	0.0196 (5)	0.0254 (6)	0.0165 (5)	0.0011 (4)	0.0073 (4)	-0.0005 (4)
C12	0.0289 (6)	0.0244 (6)	0.0284 (6)	-0.0028 (5)	0.0070 (5)	0.0001 (5)
C13	0.0364 (7)	0.0276 (6)	0.0333 (7)	0.0070 (5)	0.0096 (6)	0.0044 (5)
C14	0.0245 (6)	0.0436 (8)	0.0243 (6)	0.0088 (5)	0.0055 (5)	0.0011 (5)
C15	0.0241 (6)	0.0394 (8)	0.0292 (7)	-0.0045 (5)	0.0038 (5)	-0.0072 (6)
C16	0.0283 (6)	0.0247 (6)	0.0285 (6)	-0.0015 (5)	0.0070 (5)	-0.0028 (5)
C17	0.0220 (5)	0.0232 (5)	0.0161 (5)	0.0021 (4)	0.0070 (4)	0.0017 (4)
C18	0.0246 (6)	0.0260 (6)	0.0214 (6)	0.0039 (4)	0.0110 (5)	0.0031 (4)
C19	0.0246 (6)	0.0352 (7)	0.0295 (7)	-0.0037 (5)	0.0089 (5)	-0.0052 (5)
C20	0.0254 (6)	0.0434 (8)	0.0360 (8)	0.0026 (6)	0.0066 (6)	-0.0123 (6)
C21	0.0279 (7)	0.0770 (12)	0.0327 (8)	0.0111 (7)	0.0124 (6)	-0.0084 (7)
C22	0.0328 (7)	0.0861 (13)	0.0318 (8)	0.0097 (8)	0.0191 (7)	0.0092 (8)
C23	0.0266 (6)	0.0548 (9)	0.0276 (7)	0.0049 (6)	0.0123 (6)	0.0064 (6)
C24	0.0180 (5)	0.0374 (7)	0.0216 (6)	-0.0028 (5)	0.0070 (5)	-0.0061 (5)
C25	0.0244 (6)	0.0430 (8)	0.0245 (6)	0.0050 (5)	0.0068 (5)	0.0028 (5)
C26	0.0212 (5)	0.0312 (6)	0.0196 (5)	-0.0021 (5)	0.0063 (5)	-0.0018 (5)
C27	0.0213 (5)	0.0296 (6)	0.0189 (6)	-0.0020 (4)	0.0070 (5)	-0.0017 (4)
C28	0.0211 (5)	0.0293 (6)	0.0189 (6)	-0.0021 (4)	0.0063 (5)	-0.0004 (4)
C29	0.0214 (5)	0.0349 (7)	0.0181 (5)	0.0048 (5)	0.0081 (5)	-0.0011 (5)
C30	0.0286 (6)	0.0337 (7)	0.0373 (7)	0.0028 (5)	0.0069 (6)	-0.0039 (6)
C31	0.0389 (8)	0.0492 (10)	0.0530 (10)	-0.0079 (7)	0.0133 (8)	-0.0193 (8)
C32	0.0275 (7)	0.0824 (13)	0.0322 (8)	-0.0039 (8)	0.0056 (6)	-0.0215 (8)
C33	0.0302 (7)	0.0860 (14)	0.0261 (7)	0.0194 (8)	0.0031 (6)	0.0071 (8)
C34	0.0354 (7)	0.0438 (8)	0.0304 (7)	0.0124 (6)	0.0102 (6)	0.0085 (6)
C35	0.0228 (5)	0.0294 (6)	0.0184 (5)	-0.0013 (4)	0.0066 (5)	-0.0016 (4)
C36	0.0244 (6)	0.0373 (7)	0.0230 (6)	0.0000 (5)	0.0106 (5)	0.0010 (5)

*Geometric parameters (Å, °)*

S1—C18	1.7442 (12)	C8—C9	1.4207 (16)
S1—C17	1.7545 (12)	C9—C10	1.3755 (16)

## supplementary materials

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S2—C36	1.7370 (13)	C9—C17	1.4488 (15)
S2—C35	1.7454 (12)	C11—C16	1.3757 (17)
O1—C10	1.3517 (14)	C11—C12	1.3791 (17)
O1—C11	1.4050 (14)	C12—C13	1.3922 (18)
O2—C28	1.3512 (15)	C12—H12A	0.9500
O2—C29	1.4043 (15)	C13—C14	1.383 (2)
N1—C18	1.3452 (15)	C13—H13A	0.9500
N1—H1A	0.8800	C14—C15	1.381 (2)
N1—H1B	0.8800	C14—H14A	0.9500
N2—C18	1.3124 (16)	C15—C16	1.3844 (18)
N2—N3	1.3791 (13)	C15—H15A	0.9500
N3—C17	1.3007 (15)	C16—H16A	0.9500
N4—C8	1.3272 (15)	C19—C20	1.3892 (18)
N4—N5	1.3743 (14)	C19—C24	1.3891 (19)
N5—C10	1.3556 (15)	C19—H19A	0.9500
N5—C6	1.4288 (15)	C20—C21	1.379 (2)
N6—C36	1.3493 (17)	C20—H20A	0.9500
N6—H6A	0.8800	C21—C22	1.385 (3)
N6—H6B	0.8800	C21—H21A	0.9500
N7—C36	1.3136 (17)	C22—C23	1.387 (2)
N7—N8	1.3826 (14)	C22—H22A	0.9500
N8—C35	1.2991 (16)	C23—C24	1.3888 (19)
N9—C26	1.3279 (16)	C23—H23A	0.9500
N9—N10	1.3744 (14)	C25—C26	1.4908 (17)
N10—C28	1.3555 (15)	C25—H25A	0.9800
N10—C24	1.4279 (15)	C25—H25B	0.9800
C1—C6	1.3887 (18)	C25—H25C	0.9800
C1—C2	1.3904 (17)	C26—C27	1.4198 (17)
C1—H1C	0.9500	C27—C28	1.3787 (17)
C2—C3	1.383 (2)	C27—C35	1.4503 (16)
C2—H2B	0.9500	C29—C30	1.375 (2)
C3—C4	1.388 (2)	C29—C34	1.3781 (18)
C3—H3B	0.9500	C30—C31	1.398 (2)
C4—C5	1.3859 (19)	C30—H30A	0.9500
C4—H4B	0.9500	C31—C32	1.368 (3)
C5—C6	1.3915 (18)	C31—H31A	0.9500
C5—H5A	0.9500	C32—C33	1.369 (3)
C7—C8	1.4927 (16)	C32—H32A	0.9500
C7—H7B	0.9800	C33—C34	1.391 (2)
C7—H7C	0.9800	C33—H33A	0.9500
C7—H7D	0.9800	C34—H34A	0.9500
C18—S1—C17	86.71 (5)	C14—C15—H15A	119.9
C36—S2—C35	86.84 (6)	C16—C15—H15A	119.9
C10—O1—C11	118.17 (9)	C11—C16—C15	119.26 (12)
C28—O2—C29	118.28 (10)	C11—C16—H16A	120.4
C18—N1—H1A	120.0	C15—C16—H16A	120.4
C18—N1—H1B	120.0	N3—C17—C9	122.87 (10)
H1A—N1—H1B	120.0	N3—C17—S1	113.56 (8)
C18—N2—N3	112.82 (10)	C9—C17—S1	123.54 (9)



C17—N3—N2	113.26 (10)	N2—C18—N1	122.82 (11)
C8—N4—N5	105.94 (9)	N2—C18—S1	113.65 (9)
C10—N5—N4	110.28 (9)	N1—C18—S1	123.52 (10)
C10—N5—C6	130.08 (10)	C20—C19—C24	119.10 (13)
N4—N5—C6	119.42 (9)	C20—C19—H19A	120.4
C36—N6—H6A	120.0	C24—C19—H19A	120.4
C36—N6—H6B	120.0	C21—C20—C19	120.55 (14)
H6A—N6—H6B	120.0	C21—C20—H20A	119.7
C36—N7—N8	112.34 (10)	C19—C20—H20A	119.7
C35—N8—N7	113.19 (10)	C20—C21—C22	119.89 (13)
C26—N9—N10	105.92 (10)	C20—C21—H21A	120.1
C28—N10—N9	110.37 (10)	C22—C21—H21A	120.1
C28—N10—C24	130.33 (11)	C21—C22—C23	120.51 (15)
N9—N10—C24	118.92 (10)	C21—C22—H22A	119.7
C6—C1—C2	119.22 (12)	C23—C22—H22A	119.7
C6—C1—H1C	120.4	C22—C23—C24	119.09 (15)
C2—C1—H1C	120.4	C22—C23—H23A	120.5
C3—C2—C1	120.27 (13)	C24—C23—H23A	120.5
C3—C2—H2B	119.9	C23—C24—C19	120.84 (12)
C1—C2—H2B	119.9	C23—C24—N10	118.13 (12)
C2—C3—C4	120.04 (12)	C19—C24—N10	120.95 (11)
C2—C3—H3B	120.0	C26—C25—H25A	109.5
C4—C3—H3B	120.0	C26—C25—H25B	109.5
C5—C4—C3	120.43 (14)	H25A—C25—H25B	109.5
C5—C4—H4B	119.8	C26—C25—H25C	109.5
C3—C4—H4B	119.8	H25A—C25—H25C	109.5
C4—C5—C6	119.08 (13)	H25B—C25—H25C	109.5
C4—C5—H5A	120.5	N9—C26—C27	111.09 (11)
C6—C5—H5A	120.5	N9—C26—C25	120.35 (11)
C1—C6—C5	120.93 (11)	C27—C26—C25	128.55 (11)
C1—C6—N5	120.67 (11)	C28—C27—C26	104.34 (10)
C5—C6—N5	118.33 (11)	C28—C27—C35	127.21 (11)
C8—C7—H7B	109.5	C26—C27—C35	128.21 (11)
C8—C7—H7C	109.5	O2—C28—N10	122.78 (11)
H7B—C7—H7C	109.5	O2—C28—C27	128.84 (11)
C8—C7—H7D	109.5	N10—C28—C27	108.27 (11)
H7B—C7—H7D	109.5	C30—C29—C34	121.48 (13)
H7C—C7—H7D	109.5	C30—C29—O2	123.56 (11)
N4—C8—C9	111.05 (10)	C34—C29—O2	114.95 (12)
N4—C8—C7	120.65 (10)	C29—C30—C31	118.28 (14)
C9—C8—C7	128.29 (10)	C29—C30—H30A	120.9
C10—C9—C8	104.34 (10)	C31—C30—H30A	120.9
C10—C9—C17	127.61 (11)	C32—C31—C30	121.03 (17)
C8—C9—C17	128.04 (11)	C32—C31—H31A	119.5
O1—C10—N5	123.32 (10)	C30—C31—H31A	119.5
O1—C10—C9	128.20 (10)	C31—C32—C33	119.66 (15)
N5—C10—C9	108.38 (10)	C31—C32—H32A	120.2
C16—C11—C12	121.86 (11)	C33—C32—H32A	120.2
C16—C11—O1	114.58 (10)	C32—C33—C34	120.81 (15)

## supplementary materials

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C12—C11—O1	123.52 (11)	C32—C33—H33A	119.6
C11—C12—C13	118.12 (12)	C34—C33—H33A	119.6
C11—C12—H12A	120.9	C29—C34—C33	118.74 (15)
C13—C12—H12A	120.9	C29—C34—H34A	120.6
C14—C13—C12	120.90 (13)	C33—C34—H34A	120.6
C14—C13—H13A	119.6	N8—C35—C27	123.33 (11)
C12—C13—H13A	119.6	N8—C35—S2	113.75 (9)
C15—C14—C13	119.61 (12)	C27—C35—S2	122.91 (9)
C15—C14—H14A	120.2	N7—C36—N6	123.00 (12)
C13—C14—H14A	120.2	N7—C36—S2	113.89 (9)
C14—C15—C16	120.25 (12)	N6—C36—S2	123.06 (10)
C18—N2—N3—C17	-0.36 (15)	N3—N2—C18—S1	0.46 (14)
C8—N4—N5—C10	0.95 (13)	C17—S1—C18—N2	-0.33 (10)
C8—N4—N5—C6	-174.12 (10)	C17—S1—C18—N1	-179.62 (12)
C36—N7—N8—C35	-0.94 (17)	C24—C19—C20—C21	0.5 (2)
C26—N9—N10—C28	0.83 (14)	C19—C20—C21—C22	-0.9 (2)
C26—N9—N10—C24	-172.83 (11)	C20—C21—C22—C23	0.2 (3)
C6—C1—C2—C3	1.42 (19)	C21—C22—C23—C24	0.8 (2)
C1—C2—C3—C4	-0.8 (2)	C22—C23—C24—C19	-1.1 (2)
C2—C3—C4—C5	-0.9 (2)	C22—C23—C24—N10	175.63 (13)
C3—C4—C5—C6	2.0 (2)	C20—C19—C24—C23	0.46 (19)
C2—C1—C6—C5	-0.36 (19)	C20—C19—C24—N10	-176.23 (11)
C2—C1—C6—N5	-177.25 (11)	C28—N10—C24—C23	156.26 (13)
C4—C5—C6—C1	-1.3 (2)	N9—N10—C24—C23	-31.54 (17)
C4—C5—C6—N5	175.65 (13)	C28—N10—C24—C19	-27.0 (2)
C10—N5—C6—C1	-27.86 (19)	N9—N10—C24—C19	145.24 (12)
N4—N5—C6—C1	146.09 (11)	N10—N9—C26—C27	-0.30 (14)
C10—N5—C6—C5	155.17 (13)	N10—N9—C26—C25	178.83 (11)
N4—N5—C6—C5	-30.88 (17)	N9—C26—C27—C28	-0.31 (14)
N5—N4—C8—C9	-0.65 (13)	C25—C26—C27—C28	-179.36 (13)
N5—N4—C8—C7	178.21 (11)	N9—C26—C27—C35	174.30 (12)
N4—C8—C9—C10	0.12 (14)	C25—C26—C27—C35	-4.7 (2)
C7—C8—C9—C10	-178.63 (12)	C29—O2—C28—N10	-80.86 (15)
N4—C8—C9—C17	179.00 (11)	C29—O2—C28—C27	103.48 (15)
C7—C8—C9—C17	0.3 (2)	N9—N10—C28—O2	-177.49 (11)
C11—O1—C10—N5	-77.24 (14)	C24—N10—C28—O2	-4.8 (2)
C11—O1—C10—C9	106.74 (14)	N9—N10—C28—C27	-1.05 (14)
N4—N5—C10—O1	-177.60 (10)	C24—N10—C28—C27	171.67 (12)
C6—N5—C10—O1	-3.22 (19)	C26—C27—C28—O2	176.96 (12)
N4—N5—C10—C9	-0.90 (13)	C35—C27—C28—O2	2.3 (2)
C6—N5—C10—C9	173.49 (11)	C26—C27—C28—N10	0.81 (14)
C8—C9—C10—O1	176.97 (11)	C35—C27—C28—N10	-173.88 (12)
C17—C9—C10—O1	-1.9 (2)	C28—O2—C29—C30	-11.62 (17)
C8—C9—C10—N5	0.47 (13)	C28—O2—C29—C34	169.42 (11)
C17—C9—C10—N5	-178.42 (11)	C34—C29—C30—C31	0.5 (2)
C10—O1—C11—C16	159.41 (10)	O2—C29—C30—C31	-178.34 (13)
C10—O1—C11—C12	-22.92 (16)	C29—C30—C31—C32	0.0 (2)
C16—C11—C12—C13	0.19 (19)	C30—C31—C32—C33	-0.5 (2)
O1—C11—C12—C13	-177.32 (11)	C31—C32—C33—C34	0.4 (2)

C11—C12—C13—C14	0.3 (2)	C30—C29—C34—C33	-0.6 (2)
C12—C13—C14—C15	-0.5 (2)	O2—C29—C34—C33	178.33 (12)
C13—C14—C15—C16	0.3 (2)	C32—C33—C34—C29	0.2 (2)
C12—C11—C16—C15	-0.36 (19)	N7—N8—C35—C27	-178.05 (11)
O1—C11—C16—C15	177.35 (11)	N7—N8—C35—S2	0.50 (15)
C14—C15—C16—C11	0.1 (2)	C28—C27—C35—N8	153.50 (13)
N2—N3—C17—C9	178.14 (11)	C26—C27—C35—N8	-19.9 (2)
N2—N3—C17—S1	0.10 (13)	C28—C27—C35—S2	-24.92 (18)
C10—C9—C17—N3	159.30 (12)	C26—C27—C35—S2	161.64 (11)
C8—C9—C17—N3	-19.3 (2)	C36—S2—C35—N8	0.03 (11)
C10—C9—C17—S1	-22.85 (18)	C36—S2—C35—C27	178.59 (11)
C8—C9—C17—S1	158.51 (10)	N8—N7—C36—N6	-176.50 (13)
C18—S1—C17—N3	0.12 (10)	N8—N7—C36—S2	0.97 (15)
C18—S1—C17—C9	-177.90 (11)	C35—S2—C36—N7	-0.58 (11)
N3—N2—C18—N1	179.75 (11)	C35—S2—C36—N6	176.89 (13)

*Hydrogen-bond geometry* ( $\text{\AA}$ ,  $^\circ$ )

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
N1—H1A $\cdots$ N7	0.88	2.19	3.0266 (16)	158
N1—H1B $\cdots$ N4 <sup>i</sup>	0.88	2.21	3.0113 (15)	150
N6—H6A $\cdots$ N2	0.88	2.11	2.9467 (15)	159
N6—H6B $\cdots$ N9 <sup>ii</sup>	0.88	2.23	2.9915 (16)	145

Symmetry codes: (i)  $x, -y+3/2, z+1/2$ ; (ii)  $x, -y+3/2, z-1/2$ .

Fig. 1

